

# Carbon Cycle

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## What is a Biogeochemical Cycle?

*“Biogeochemical cycles mainly refer to the movement of nutrients and other elements between biotic and abiotic factors.”*

The term biogeochemical is derived from “**bio**” meaning **biosphere**, “**geo**” meaning the **geological components** and “**chemical**” meaning the **elements that move through a cycle**.

## Carbon Cycle

The **Carbon Cycle** is Earth's natural process for recycling carbon atoms, which are fundamental to all life. Carbon continuously moves between the atmosphere, oceans, land (living organisms and soil), and the Earth's crust (rocks and fossil fuels). This complex biogeochemical cycle involves several key processes:

**Reservoirs:** Carbon is stored in several major reservoirs:

- **Atmosphere:** Primarily as carbon dioxide (CO<sub>2</sub>).
- **Oceans:** Dissolved CO<sub>2</sub>, carbonic acid, bicarbonate, and carbonate ions, as well as in marine organisms and sediments.
- **Land (Terrestrial):** In plants (biomass), soils (organic matter), and rocks (e.g., limestone).
- **Fossil Fuels:** Stored carbon from ancient organic matter (coal, oil, natural gas).

## Key Processes:

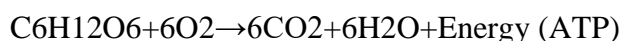
1. **Photosynthesis:** Plants and other photosynthetic organisms absorb carbon dioxide (CO<sub>2</sub>) from the atmosphere (or dissolved in water) to produce organic compounds (sugars) for growth, releasing oxygen. This removes carbon from the atmosphere.



Where:

- CO<sub>2</sub> is carbon dioxide
- H<sub>2</sub>O is water
- C<sub>6</sub>H<sub>12</sub>O<sub>6</sub> is glucose (a sugar)
- O<sub>2</sub> is oxygen

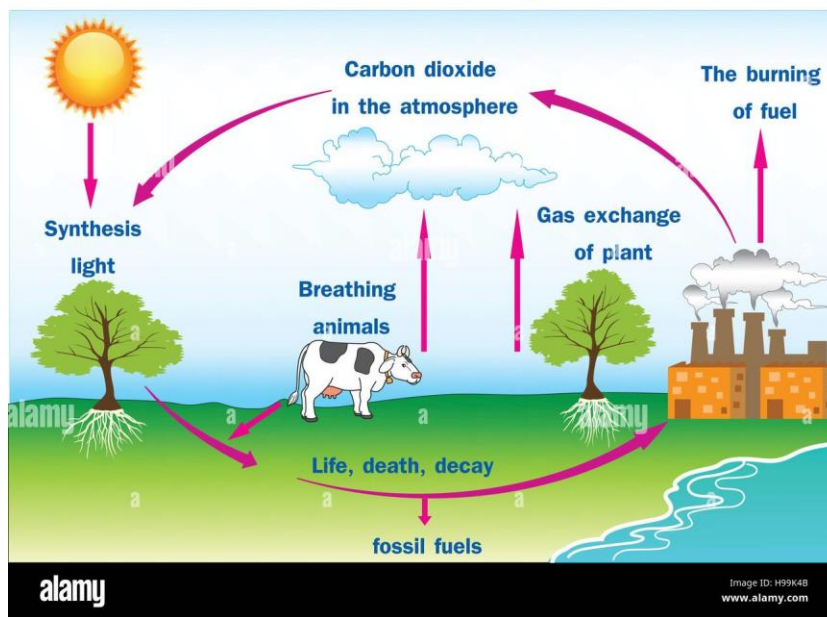
2. **Respiration:** All living organisms (plants, animals, microbes) release CO<sub>2</sub> back into the atmosphere through cellular respiration as they break down organic matter for energy.



Where:

- $C_6H_{12}O_6$  is **Glucose** (a sugar, the primary energy source)
- $O_2$  is **Oxygen**
- $CO_2$  is **Carbon Dioxide**
- $H_2O$  is **Water**
- **ATP** is **Adenosine Triphosphate**, the main energy currency of the cell. While the equation often just shows "Energy" or "ATP," it's understood that a significant amount of ATP (around 30-32 molecules) is produced.

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3. **Decomposition:** When plants and animals die, decomposers (bacteria and fungi) break down organic matter, releasing carbon back into the soil and atmosphere as  $CO_2$  or methane ( $CH_4$ ).
4. **Combustion:** The burning of organic matter, such as wood, or fossil fuels (coal, oil, natural gas) formed from ancient organic remains, releases large amounts of  $CO_2$  into the atmosphere.
5. **Oceanic Exchange:** Carbon is constantly exchanged between the atmosphere and the ocean.  $CO_2$  dissolves in surface waters, and marine organisms use it to build shells and skeletons. Carbon can then be stored in deep ocean currents or as sediments over long periods.
6. **Sedimentation and Geologic Processes:** Over millions of years, dead organisms and carbon-containing sediments can be buried and compressed, forming rocks like limestone and fossil fuels. Volcanic activity can release this stored carbon back into the atmosphere.

### Significance:

The carbon cycle is incredibly significant for life on Earth for several reasons:

- **Foundation of Life:** Carbon is the backbone of all organic molecules (proteins, DNA, carbohydrates, fats) that make up living organisms. The carbon cycle ensures a continuous supply of this essential element.
- **Climate Regulation:** Carbon dioxide is a crucial greenhouse gas. It traps heat in the Earth's atmosphere, preventing the planet from becoming a frozen, uninhabitable ice ball. A balanced carbon cycle helps maintain Earth's temperature within a range suitable for life.
- **Energy Flow:** The cycle facilitates the transfer of energy through ecosystems. Photosynthesis converts solar energy into chemical energy stored in carbon compounds, which then moves through food webs as organisms consume each other.
- **Nutrient Cycling:** It's intrinsically linked with other biogeochemical cycles (like the nitrogen and water cycles), influencing soil fertility and overall ecosystem health.
- **Ocean Health:** The ocean plays a vital role in absorbing atmospheric CO<sub>2</sub>. However, increased atmospheric CO<sub>2</sub> due to human activities leads to ocean acidification, which threatens marine life, especially organisms with calcium carbonate shells